

225(4): Effect of Weak Field on Atoms and Molecules
Spectra.

Consider the general equation in the form:

$$\hat{E}_3 \phi^R = H_3 \phi^R \quad - (1)$$

$$\hat{E}_3 \phi^L = H_3 \phi^L \quad - (2)$$

In order to consider the weak nuclear interaction the left hand spinors is extended to:

$$\phi^L \rightarrow \begin{bmatrix} \phi^L \\ \sim \end{bmatrix} - (3)$$

but the right hand spinors is unaffected. So eq. (2)

$$\hat{E}_3 \begin{bmatrix} \phi^L \\ \sim \end{bmatrix} = H_3 \begin{bmatrix} \phi_L \\ \sim \end{bmatrix} - (4)$$

For the electron part of this equation:

$$\hat{E}_3 \phi^L = \frac{1}{2m} \underline{\sigma} \cdot \underline{\Pi} \underline{\sigma} \cdot \underline{\Pi} - (5)$$

where $\underline{\Pi} = \underline{p} - e\underline{A} - g \frac{\underline{w}}{m}$ - (6)
 is the rotation of previous notes. Therefore m is
 the mass of the electron

1). For real $\underline{\pi}$:

$$\underline{\sigma} \cdot \underline{\pi} \underline{\sigma} \cdot \underline{\pi} = \underline{\pi} \cdot \underline{\pi} + i \underline{\sigma} \cdot \underline{\pi} \times \underline{\pi} - (7)$$

So

$$\begin{aligned}\underline{\sigma} \cdot \underline{\pi} \underline{\sigma} \cdot \underline{\pi} &= \underline{\sigma} \cdot (\underline{p} - e\underline{A} - g\underline{N}) \underline{\sigma} \cdot (\underline{p} - e\underline{A} - g\underline{N}) \\&= \underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{p} - e \underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{A} - g \underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{N} \\&\quad - e \underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{p} + e^2 \underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{A} + eg \underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{N} \\&\quad - g \underline{\sigma} \cdot \underline{N} \underline{\sigma} \cdot \underline{p} + ge \underline{\sigma} \cdot \underline{N} \underline{\sigma} \cdot \underline{A} + g^2 \underline{\sigma} \cdot \underline{N} \underline{\sigma} \cdot \underline{N}\end{aligned} - (8)$$

Here:

$$\underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{A} = \underline{p} \cdot \underline{A} + i \underline{\sigma} \cdot \underline{p} \times \underline{A} - (9)$$

$$\underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{p} = \underline{A} \cdot \underline{p} + i \underline{\sigma} \cdot \underline{A} \times \underline{p} - (10)$$

$$\underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{N} = \underline{p} \cdot \underline{N} + i \underline{\sigma} \cdot \underline{p} \times \underline{N} - (11)$$

$$\underline{\sigma} \cdot \underline{N} \underline{\sigma} \cdot \underline{p} = \underline{N} \cdot \underline{p} + i \underline{\sigma} \cdot \underline{N} \times \underline{p} - (12)$$

$$\underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{N} = \underline{A} \cdot \underline{N} + i \underline{\sigma} \cdot \underline{A} \times \underline{N} - (13)$$

$$\underline{\sigma} \cdot \underline{N} \underline{\sigma} \cdot \underline{A} = \underline{N} \cdot \underline{A} + i \underline{\sigma} \cdot \underline{N} \times \underline{A} - (14)$$

As in note 225(2) there are terms such as:

$$\begin{aligned}
 \hat{H}_w \phi^L &= -\frac{i\hbar}{2m} \left(\underline{\sigma} \cdot \underline{w} \times \hat{\underline{p}} + \underline{\sigma} \cdot \hat{\underline{p}} \times \underline{w} \right) \phi^L \\
 &= \frac{e\hbar g}{2m} (\underline{\sigma} \cdot (\underline{w} \times \underline{\nabla} + \underline{\nabla} \times \underline{w})) \phi^L \\
 &= -\frac{g\hbar e}{2m} (\underline{\sigma} \cdot \underline{\nabla} \times \underline{w}) \phi^L \quad -(7)
 \end{aligned}$$

So the energy for left handed electron is:

$$E_L = -\frac{\hbar}{2m} (e\underline{\sigma} \cdot \underline{B} + g\underline{\sigma} \cdot \underline{\nabla} \times \underline{w}) \quad -(8)$$

For right handed electron:

$$E_R = -\frac{\hbar e}{2m} \underline{\sigma} \cdot \underline{B} \quad -(9)$$

This means that the following effects are slightly different for right and left handed electrons:

Faraday effect, Zeeman effect, ESR, NMR, FMR, spin-orbit coupling, Thomas precession, Diamagnetic term, and similar.
