

251(2): Expectation Value of Energy

In the previous note the Hamiltonian was derived in the format:

$$\hat{H}\psi = -\frac{e\hbar}{2m} \underline{\sigma} \cdot \underline{B}_1 \left(\psi + r \frac{\partial \psi}{\partial r} \right) + \frac{e}{2m} \underline{\sigma} \cdot \underline{B}_1 \underline{\sigma} \cdot \underline{L} \psi \quad (1)$$

where:

$$\underline{B}_1 = \underline{B} - \underline{e}_r (\underline{e}_r \cdot \underline{B}) \quad (2)$$

and

$$\underline{e}_r = \frac{\underline{r}}{r} \quad (3)$$

Therefore there are three types of Hamiltonian:

$$\hat{H}_1 \psi = -\frac{e\hbar}{2m} \underline{\sigma} \cdot \underline{B}_1 \psi \quad (4)$$

$$\hat{H}_2 \psi = -\frac{e\hbar}{2m} \underline{\sigma} \cdot \underline{B}_1 r \frac{\partial \psi}{\partial r} \quad (5)$$

$$\hat{H}_3 \psi = \frac{e}{2m} \underline{\sigma} \cdot \underline{B}_1 \underline{\sigma} \cdot \underline{L} \psi \quad (6)$$

The energy expectation values are:

$$E_1 = -\frac{e\hbar}{2m} \int \psi^* \underline{\sigma} \cdot \underline{B}_1 \psi d\tau \quad (7)$$

$$E_2 = -\frac{e\hbar}{2m} \int \psi^* \underline{\sigma} \cdot \underline{B}_1 r \frac{\partial \psi}{\partial r} d\tau \quad (8)$$

$$E_3 = \frac{e}{2m} \int \psi^* \underline{\sigma} \cdot \underline{B}_1 \underline{\sigma} \cdot \underline{L} \psi d\tau \quad (9)$$

2) so there are two different types of resonance spectra. In
note 251(1) it evaluation was initiated of type (a)
with hydrogenic wave functions.

The eigenfunctions of type (1) contain the
conventional ESR Hamiltonian:

$$\hat{H}_{\text{ESR}} \psi = - \frac{e\hbar}{2m} \underline{\sigma} \cdot \underline{B} \psi \quad (10)$$

The next note will evaluate some of the eigenvalues
by hand for the simplest hydrogenic wave functions.
In general, all three Hamiltonians will be affected by
the chemical shift.

Therefore a considerable amount of new information
has emerged from the well known:

$$\hat{H} \psi = - \frac{e}{2m} \left(\underline{\sigma} \cdot \underline{p} \underline{\sigma} \cdot \underline{A} + \underline{\sigma} \cdot \underline{A} \underline{\sigma} \cdot \underline{p} \right) \psi \quad (11)$$
